A METHOD FOR DETERMINING ABSOLUTE CONFIGURATION OF $\beta\mbox{-}{\mbox{aMINO}}$ ACIDS BY CD SPECTRA OF THEIR DNP DERIVATIVES -

Ukon Nagai*, Masao Kawai,

Mitsubishi-Kasei Institute of Life Sciences,

11-Minamiooya, Machida-shi, Tokyo, 194

Takashi Yamada, Shigeru Kuwata, and Hiroshi Watanabe
Faculty of Sciences, Konan University, Okamoto, Higashinada-ku, Kobe, 658

Summary: Aromatic β -amino acids were converted to the Dnp derivatives, and aliphatic ones to Dnp-pMA derivatives. The sign of the Cotton effect near 400 nm reflected the configuration at the β -position without exception. Thus, the relation seems to afford a new reliable method for determining the absolute configuration of β -amino acids.

N-2,4-Dinitrophenyl(Dnp) derivatives of aromatic α -amino acids exhibit characteristic CD spectra above 300 nm. The sign of the band near 400 nm has high degree of correlation to the absolute configuration at the α -carbon atom (Dnp-aromatic rule). The characteristic CD pattern is considered to be caused by interaction of the Dnp- and the aromatic chromophores. CD spectra of similar pattern but of opposite sign were observed for p-methoxyanilides(pMA) of aliphatic Dnp- α -amino acids with the same configuration. In this paper, extension of the two rules to β -amino acids is described.

Some antibiotic peptides, such as bottromycin, bleomycin, iturin A, blasticidin S, and so on, contain β -amino acids as their components. Since usual enzymatic methods cannot be applied to β -amino acids, chiroptical methods are precious for determining their configuration. Only a few papers appeared in literature concerning the chiroptical relations of β -amino acid derivatives, β -and the relations are not always satisfactory for determining the configuration of various β -amino acids because some exceptional cases were pointed out in each case. The method to measure the CD spectra of dicyclohexylammonium(DCHA) salts of N-dithiocarbamoyl derivatives in CHCl₃ or benzene seems to be the most reliable so far reported. β

For the purpose of offering a new method for determining the absolute configuration of β -amino acids and of confirming the scope of the Dnp-aromatic rule in its generalized form, 9) CD spectra of a number of Dnp- β -amino acids and their pMA derivatives were measured.

Figs. 1 and 2 reproduce CD spectra of the Dnp-derivatives of five aromatic β -amino acids and Dnp-pMA derivatives of five aliphatic β -amino acids, respectively. The CD curves indicate that the same rule as for α -amino acids are applicable also to β -amino acid derivatives: i.e., in the case of L-series compounds the Cotton effect of the band near 400 nm has negative sign for the Dnp-derivatives of aromatic amino acids, and positive sign for the Dnp-pMA derivatives of aliphatic amino acids. The relation will be useful for determining the stereochemistry of newly found β -amino acids.

In fact, the stereochemistry of an iturinic acid was assigned by this method, 10) and the

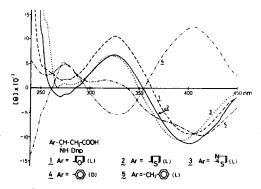


Fig. 1: CD Spectra of Dnp-derivatives of aromatic β -amino acids

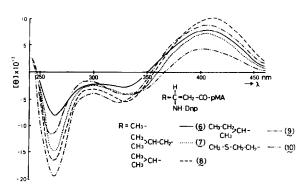


Fig. 2: CD Spectra of Dnp-L- β -aminoacyl-pMA

assignment has been further confirmed by the present results. Moreover, the rule was applied to a β -amino acid obtained from the hydrolysate of tallysomycin, ¹¹⁾ β -amino- β -(4-amino-6-carboxy-5-methylpyrimidin-2-yl)propionic acid (11), whose configuration was not determined yet due to the scarcity of the sample. Its Dnp-derivative showed positive Cotton effect near 400 nm ([θ] = +910) indicating D-configuration at the β -carbon atom. The same compound isolated from bleomycin has also D-configuration. The low [θ] max -value would probably be due to partial recemization during hydrolysis.

Dinitrophenylation was carried out according to the Sanger's procedure, 12) and pMA derivatives were prepared by treating the Dnp-amino acids dissolved in $\mathrm{CH_2Cl_2}$ with 1-ethyl-3-(3-dimethylaminopropyl)-carbodimide, 1-hydroxybenzotriazole, and p-anisidine. 13) The purity and structure of each compound were confirmed by TLC, 1 H-NMR, and elemental analyses except 11 , which could not be purified completely.

<u>Acknowledgement</u>: The authors are grateful to Dr. Konishi of Bristol-Banyu Research Institute Ltd. for the generous gift of a precious sample of compound 11.

References

- 1) M.Kawai, U.Nagai, and M.Katsumi, Tetrahedron Letters, 2845-2848 (1975)
- 2) M.Kawai and U.Nagai, ibid., 3889-3890 (1977)
- 3) B.Sjöberg, B.Hansson, and R.Dahlbom, Acta Chem. Scand., 16, 1057-1059 (1962)
- 4) H.Yonehara and N.Otake, Tetrahedron Letters, 3785-3791 (1966)
- 5) Y.Seto, T.Yamada, K.Niwa, S.Miwa, F.Tanaka, S.Kuwata, and H.Watanabe, Chem. Lett., 151-154 (1973)
- 6) T.Yamada, S.Kuwata, and H.Watanabe, <u>Tetrahedron Letters</u>, 1813-1816 (1978); S.Kuwata, T.Yamada, T.Shinogi, N.Yamagami, F.Kitabashi, T.Miyazawa and H.Watanabe, <u>Bull. Chem. Soc. Jpn.</u>, <u>52</u>, 3326-3328 (1979)
- 7) N.Nakamura, T.Yoshioka, T.Takita, H.Umezawa, Y.Muraoka, and Y.Iitaka, <u>J</u>. <u>Antibiot</u>., <u>29</u>, 762-764 (1976)
- 8) J.M.Cassel, A.Fürst, and W.Meier, <u>Helv</u>. <u>Chim</u>. <u>Acta</u>, <u>59</u>, 1917-1924 (1976)
- 9) M.Kawai, U.Nagai, M.Katsumi, and A.Tanaka, Tetrahedron, 34, 3435-3444 (1978)
- 10) U.Nagai, F.Besson, and F.Peypoux, Tetrahedron Letters, 2359-2360 (1979)
- 11) M.Konishi, K.Saito, K.Numata, T.Tsuno, K.Asama, H.Tsukiura, T.Naito, and H.Kawaguchi, J. Antibiot., 30, 789-805 (1977)
- 12) F.Sanger, Biochem. J., 39, 507-515 (1945)
- J.C.Sheehan, J.Preston, and P.A. Cruickshunk, J. Am. Chem. Soc., 87, 2492-2493 (1965);
 S.Nozaki, S.Kimura, and I.Muramatsu, Chem. Lett., 1057-1058 (1979)